

A FORM OF $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase OF HUMAN RED CELL MEMBRANES WITH LOW AFFINITY FOR Mg-ATP: A HYPOTHESIS FOR ITS FUNCTION

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We have shown before that the cell membrane is responsible for the entire true ATPase activity of human red cells and that in the absence of membranes a steady state of glycolysis cannot exist [1]. Thus the ATP consumption by membrane ATPases is the main controlling factor of red cell glycolysis and is equal to the ATP production which amounts to ~ 2 mmol ATP per liter cells and hour at pH 7.2 and 37°C .

The known ion transport ATPase activities amount to $\lesssim 25\%$ of the total ATP consumption. Most of it is due to $(\text{Na}^+ + \text{K}^+ + \text{Mg}^{2+})$ -ATPase activity, the share of 20% of which is indicated by the effect of ouabain on red cell glycolysis [2] and on the total ATP-breakdown of glucose-free cells (fig.1). It is even smaller at low pH-values and temperatures [3].

The extent of Ca^{2+} transport under physiological conditions is very small indeed, owing to the low passive permeability of the mature human red cell for Ca^{2+} . Only ~ 1 μmol calcium per liter cells and hour enters the cells at 37°C [4]. The low permeability of the red cell membrane to Ca^{2+} is changed little during cellular ATP depletion [5]. Thus $<1\%$ of the ATP production is used for Ca^{2+} transport in intact red cells.

The correctness of this assessment is borne out by the negligible effect of lanthanum ions, the most effective inhibitor of active Ca^{2+} -transport in human red cells [6], on the ATP-breakdown of glucose-free cells (fig.1).

Investigations on the ATP depletion of glucose-free cells with and without ouabain demonstrated that the rate of ATP-breakdown declined with lower ATP concentrations ([3] and fig.1). These results indicated the presence of an ATPase with low affinity for Mg-ATP in intact red cells (~ 1 mM Mg-ATP). Therefore we began a search for an ATPase with such characteris-

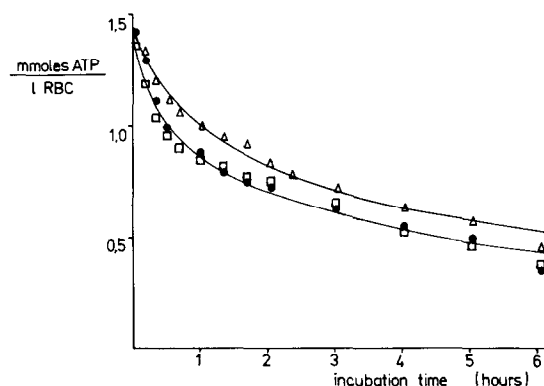


Fig.1. ATP-breakdown in intact glucose-free human red blood cells at 37°C incubated in a medium containing 145 mM NaCl, 5 mM KCl, 1 mM MgCl_2 , 1 mM Na_2HPO_4 , 1.5 mM CaCl_2 and 10 mM imidazole at pH 7.4. Red cells from fresh, heparinized blood were washed twice with glucose (5 mM) containing solution (as above) at 37°C to maintain the steady state concentrations of metabolites. The red cells were depleted of glucose by two further washing steps without glucose at 37°C and incubated at hematocrit values of 42%. (●—●) Control; (△—△) addition of 0.2 mM ouabain or 0.2 mM ouabain + 0.1 mM La^{3+} ; (□—□) addition of 0.1 mM La^{3+} .

tics. Fig.2 demonstrates that the $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase activity of 'isotonic' membranes has $K_m \sim 0.76$ mM Mg-ATP, in contrast to a 'hypotonic' membrane preparation, which had a 20-fold higher affinity (K_m 0.035 mM). Washing of 'isotonic' membranes with a hypotonic solution resulted in a 5-fold increase of affinity for Mg-ATP.

The K_m -value of Mg-ATP of the 'hypotonic' preparation corresponds to that of the 'B-state' of the $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase as in [7]. The data presented indicate that there exists a factor in red cells in addi-

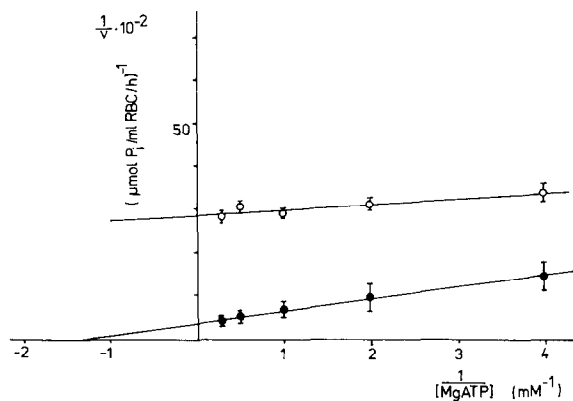


Fig. 2. Double reciprocal plot of $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase activity as a function of Mg-ATP concentration in media with 1 mM Mg^{2+} . The reaction mixture contained 120 mM KCl, 20 mM NaCl, 1 mM dithioerythritol, 0.2 mM ouabain and 10 mM imidazole (pH 7.2). Membranes were assayed for ATPase activity by incubation at 37°C for 15 min and the P_i liberated was determined by the method in [12]. The ATPase activities of the membranes were normalized to 1 ml red blood cells in relation to the recovery of acetylcholinesterase activity [13] of the whole hemolysate. (○—○) 'Hypotonic' membranes, prepared under hypotonic conditions in 10 mM imidazole (pH 7.2) and freeze-thawed for ATPase determination. (●—●) 'Isotonic' membranes; prepared in a solution containing 120 mM KCl, 4 mM 2-mercaptoethanol, 10 mM imidazole (pH 7.2), 1 mM MgEt_2 , 20 mM NaCl. Hemolysis was carried out with a 10% cell suspension by a freeze-thaw procedure or by addition of 0.02% saponin.

tion to calmodulin, which modulates the properties of $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase in red cells.

Calmodulin is known to increase the affinity for Ca^{2+} and the maximal activity of the enzyme [8], while the affinity for Mg-ATP may not be changed greatly [9]. The additional factor apparently affects primarily the affinity for Mg-ATP and maintains the high $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase activity of isotonically isolated red blood cell membranes. It is possible that this postulated factor is identical with the modulator-binding protein found in brain, which is reported to antagonize the $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase activation of red blood cell membranes by calmodulin [10,11].

We assume that $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase, calmodulin and the modulator-binding protein constitute a system by which the affinity and activity of this ATPase is adjusted over a wide range. Presumably the modulation of the affinity for Mg-ATP is of greater importance than that for Ca^{2+} in intact red cells. Furthermore we should like to propose that this

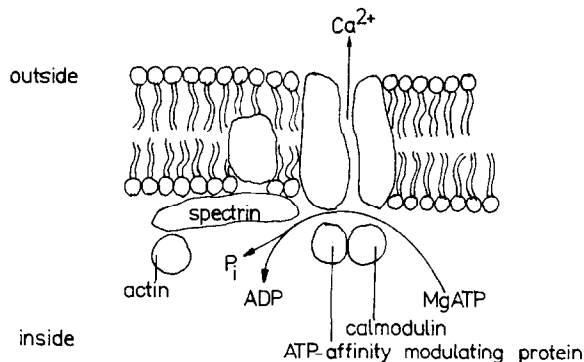


Fig. 3. Hypothetical model for the action of red blood cell membrane $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase at the intracellular $[\text{Ca}^{2+}]$ (probably 1 μM). A Mg-ATP affinity modulating protein is assumed to form a complex with Ca-ATPase and calmodulin which are bound to the membrane. The following properties are postulated: (1) At low $[\text{Ca}^{2+}]$ the complex is assumed to have both Mg-ATP-dependent kinase activity and Ca^{2+} -dependent phosphatase activity with low Mg-ATP affinity which phosphorylate and dephosphorylate so far unspecified membrane proteins; (2) Increase of intracellular $[\text{Ca}^{2+}]$ dissociates the modulating protein from the complex so that $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase + calmodulin acts as a Ca^{2+} -transporting system with high ATP affinity.

system has functions which are not limited to Ca^{2+} transport through the cell membrane.

We suggest that its function is a combination of a kinase reaction dependent on Mg-ATP with a phosphatase reaction dependent on Ca^{2+} , which might result in an apparent $(\text{Ca}^{2+} + \text{Mg}^{2+})$ -ATPase activity separate from the Ca^{2+} -pump and which affects the structure and dynamics of the cell membrane. We assume that this system is responsible for the bulk of ATP consumption of the red cell membrane. A scheme of our hypothesis is shown in fig. 3.

References

- [1] Maretzki, D., Brenneis, M., Schwarz, Zs., Lange, I. and Rapoport, S. (1977) *Acta Biol. Med. Germ.* 36, 625–629.
- [2] Whittam, R. and Ager, M. E. (1965) *J. Biochem.* 97, 214–227.
- [3] Tu, L. D., Maretzki, D., Rapoport, I., Elsner, R. and Rapoport, S. (1979) *Acta Biol. Med. Germ.* 38, in press; Rapoport, I., Rapoport, S., Maretzki, D. and Elsner, R. (1979) *Acta Biol. Med. Germ.* 38, in press.
- [4] Palek, J., Church, D. and Fairbanks, G. (1976) in: *Membranes and Diseases* (Leaf, A. et al. eds) pp. 41–60, Raven Press, New York.

- [5] Schatzmann, H. J. and Vincenzi, F. F. (1969) *J. Physiol.* London 231, 369–395.
- [6] Sarkadi, B., Szász, I., Gerlezy, A. and Gardos, G. (1977) *Biochim. Biophys. Acta* 464, 93–107.
- [7] Scharff, O. (1978) *Eur. Red Cell Club Meet.* Sandbjerg, Denmark, abst.
- [8] Vincenzi, F. F. and Farrance, M. L. (1977) *J. Supramol. Struct.* 7, 301–306.
- [9] Klinger, R. (1979) 5th Joint Symp. Biochem. Soc. USSR-GDR, Weimar, abst.
- [10] Larsen, F. L., Raess, B. U., Hinds, T. R. and Vincenzi, F. F. (1978) *J. Supramol. Struct.* 9, 269–274.
- [11] Lynch, T. J. and Cheunig, W. Y. (1979) *Arch. Biochem. Biophys.* 194, 165–170.
- [12] Taussky, H. M. and Shorr, E. (1953) *J. Biol. Chem.* 202, 675–685.
- [13] Ellmann, G. L., Courtney, K. D., Andres, V. and Featherstone, R. M. (1961) *Biochem. Pharmacol.* 7, 88–95.